SAXS STUDY OF ISOTHERMAL SINTERING OF SILICA AEROGEL

9

Thierry LOURS and Jerzy ZARZYCKI

are

Aldo F. CRAIEVICH Laboratory of Science of Vitreous Materials, (LA 1119), University of Montpellier II (France)

Sao Paulo, Brazil Laboratorio Nacional de Luz Sincrotron/CNPq, Campinas (SP) and Instituto de Fisica/USP,

ther the

Michel A. AEGERTER and Dayse I. DOS SANTOS

ínstituto de Fisica e Quimica de Sao Carlos, Universidade de Sao Paulo, Sao Carlos, SP (Brazil)

Isothermal sintering of SiO<sub>2</sub> aerogels was followed by small angle X-ray scattering using synchrotron radiation. An improved treatment of SAXS data which takes into account the presence of fluctuations explains the discrepancy previously observed with He-pycnometric methods. The matrix density of the gel is shown to increase in several steps, first by a diffusion-controlled mechanism and then by viscous flow.

ishlowed, at higher temperatures (1000-1200°C) ndergoes a diffusion-controlled contraction, temperatures (500°-800°C.) the dry gel first occurence of two distinct stages : at lower because of the importance of this process in the teen the object of numerous investigations synthesis of glasses from precursors 1-2 Dilatometric studies seem to indicate the The densification of silica gels has recently

1.7-2.0 g/cm³ range ined by the He-pycnometry which are in the inlace  $0.122.0~\text{n/cm}^3$  rance 9-11contradiction with skeletal density values ob-<sup>©Servationso</sup>. These results are, however, in results of independent electron microscopy (TEM) of the order of 0.3-0.4 g/cm3 confirmed the censity of the gel matrix is quite small. Values  $^{
m kattering}$  (SAXS) have shown $^{4-7}$  that the initial Preliminary studies using small angle X-ray

can be explained by the fact that the densities "I will be shown that the previous discrepancy MAXS data is applied to evaluate the results of Softhermal sintering studies of SiO<sub>2</sub> aerogels. in the present study an improved treatment of

SAMPLE PREPARATION

to the same particle scale.

obtained by different methods are not relative

flow of oxygen. oxidised at 500°C. for 5 hours under controlled Tc = 300°C); the resulting aerogels were hypercritical conditions ( $P_c = 200 \text{ bar}$ , molar ratio  $H_20$ : TMOS = 4. Sol formation was for one week. The wet aged gels were dried in hermetically closed Pyrex tubes kept at 55°C. After vigorous stirring, gelation occurred in catalysed by  $HNO_3$  with pH = 2. solution bidistilled water was added in the tetramethoxysilane (TMOS) dissolved in methanol. The composition was 50 vol% of methanol. To this Silica gels were prepared from solutions of

rent density  $(
ho_{f a})$  was obtained from Hg-volumetry. using Micromeritics 2100E equipment, the appaorder to have identical initial characteristics the samples were taken from the same batch in conditions at 615°, 780°, 915° and 1060°C. All The nitrogen BET specific area(s) was determined Densification was studied in isothermal 1. INTRODUCTION

y viscous flow sintering

Samples destined for SAXS measurements were cut in the form of thin slices of optimum thickness and polished on both faces.

## 3. SAXS METHOD

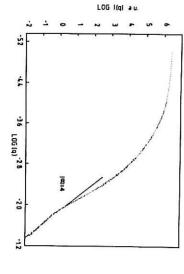
SAXS measurements were obtained at LURE synchrotron radiation facility (Orsay, France) with  $\lambda$  = 1.55 Å. The point-like cross-section of the beam dispenses from slit-height desmearing procedures. Experimental details have already been described elsewhere  $^4$ .

The SAXS intensity I(q) vs.modulus of the scattering vector q were obtained on a relative scale. A two-level electronic density model was used to evaluate the SAXS results. The low-q region was interpreted using GUINIER's approximation

$$I(q)=I_0\exp[-\frac{1}{3}\langle R_q \rangle^2 q^2]$$

where  $\langle R_{g} \rangle$  is the average gyration radius and  $I_{0}$  a constant.

In the high-q region positive departures from POROD's law were observed. The log I(q) vs. log q plots have systematically a slope greater than 4 and slowly approach 4 for the highest values of q (Fig.1).



Typical Log I(q) vs.log q plot of SAXS intensity of a silica aerogel.

Therefore RULAND's treatment  $^{12}\ \mbox{was adopted $h_{\rm P,e}$}$  with

$$I(q) = \frac{A}{q^4} + B$$

where B is a measure of density fluctuations. The constants A and B were obtained from  $I(q)_q^4$  vs.q<sup>4</sup> plot (Fig.2).

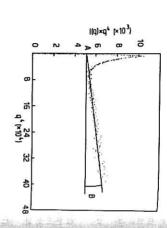


FIGURE 2  $(q)q^4vs.q^4$  plot showing the influence of the fluctuation term used to calculate the A and B constants (see text).

The volume fractions  $\varphi$  and 1- $\varphi$  of the two phase were obtained from the expression :

$$\frac{A}{Q_0} = \frac{\delta \rho_a}{\pi \phi (1-\phi)}$$

where the term  $Q_0$  is the integrated intensity

$$Q_0 = \int I(q)q^2 dq$$

numerically calculated from the experimental curve.

## RESULTS

The initial characteristics of the oxidised aerogel are : S = 415 m²/g,  $\rho_a$  = 0,42 g/cm³ and  $\langle R_{\rm q} \rangle$  = 62,3 Å.

Table I shows the evolution of S and  $\rho_{a}$  as a function of time for various temperatures of

 $_{0}$  treatment, as well as the calculated value of  $\langle R_{g} \rangle$  and the matrix density  $\rho = \rho_{a}/(1-\varphi)$  in the hypothesis of scattering by pores of volume fraction  $\varphi$  embedded in a (light) matrix of volume fraction  $1-\varphi$ 

For temperatures lower than 915°C. the plots of  $\log |\rho(t)-\rho_o|/\rho_o$  vs.log t where the initial strix density  $\rho_o$  is close to 0.52 g/cm  $^3$  are sensibly linear with a slope close to 0.46 (Fig.3). This is indicative of a diffusion-controlled mechanism involving fluctuations and ranoporosity which explains the previous deviation from POROD's law. The variations of  $\langle R_g \rangle$  are small.

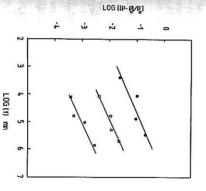


FIGURE 3 Polative increase of matrix density with time during thermal treatments at x 615°C, o  $780^{\circ}$ C, 915°C.

At 915°C. the decrease of  $< R_g >$  and of the 20°Te fraction  $\phi$  to 0.5 as well as an increase of  $\rho$  indicate that densification occurs within the 2 Tatrix (decay of nanoporosity) and that mesopocosity starts to be eliminated.

<sup>a</sup> diffusion-controlled mechanism still seems appropriate for this stage.

For temperatures above 1000°C, the fluctua-

nano-heterogeneities of the matrix are completely eliminated.

Progressive densification of the matrix close to theoretical value is achieved  $^5$ . The variations of  $\langle R_j \rangle$  with time show a coalescencetypical behaviour for the pores. The increasing q-domain where PORDD's law is obeyed indicates an increase of the length scale where the matrix is homogeneous.

A correlation of the results obtained here with results already published  $^5$  shows a monotonic variation of  $\rho$  vs.  $\rho_a(\text{Fig.4})$ .

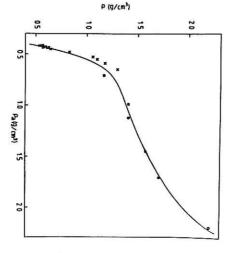


FIGURE 4 Evolution of matrix density  $\rho$  versus apparent density  $\rho_{\rm d}$  during sintering x this work, o Ref(5).

The two distinct slopes of the curve clearly indicate two densification regimes : at first, densification of the matrix without much increase of apparent density, then simultaneous increase of  $\rho$  and  $\rho_a$  which point to the elimination of meso- and macro-porosity.

The apparent discrepancy between various methods of evaluation of  $\rho$  is thus readly explained : the light matrix values observed by

Nort

TABLE 1 : Evolution of microstructural characteristics for various heat treatments.

* the values	1060	915	780	615	Temperature (°C)
of of	15 25 30	30 60 135 235	60 120 200 290	60 120 150	time (mn)
A C	314 248 225 212	329 325 331 296	382 372 376 374 370	360 365 386	S (m²/g)
are greater than 0.25	0.59 0.88 1.00 1.33	0.48 0.55 0.53 0.65	0.43 0.43 0.44 0.44	0.42 0.42 0.44	$\rho_{a}$ (g/cm <sup>3</sup> )
than 0.25	59.4 55 54 55.45 66.15	53 51 52	59.3 61 61.1 62.8	57.6 56.2	<r> (Å)</r>
	5.5 1.41 0.42 0.42	5.19 5.51 3.3	1.91 2.19 2.24 3.29 4.3	1.68 2.75 2.06	B × 10 <sup>2</sup>
	not calculated*	0.83 1.1 1.06 1.3	0.583 0.59 0.618 0.621 0.648	0.558 0.560 0.570	ρ (g/cm³)

SAXS correspond to particles 30-100 Å containing (open) nanoporosity. This can be penetrated in He-pycnometry but is seen by SAXS only as fluctuation residues. The  $\rho$ -values obtained by SAXS and those by He-pycnometry progressively coincide at later densification stages after elimination of nanoporosity.

## REFERENCES

- J. Zarzycki "Processing of gel Glasses", in Glass Science and Technology, Vol.2,p.209-249 D.R. Uhlmann, N.J. Kreidl eds., Academic Press (1984)
- 2. Proc.IV. Intern.Workshop.on glasses and glass Ceramics from gels. Kyoto 13-15 july 1987, J.Non-Cryst.Sol. 100,(1988) 1-554.
- 3. Prassas M., Phalippou J., Zarzycki J. in "Science of Ceramic Chemical processing" L.L. Hench, D.R.Ulrich Eds., Wiley(1986),p.156.

  4. A.F. Craievich, M.A. Aegerter, D.I. Dos Santos T. Woignier, and J.Zarzycki.
  J.of Non-Cryst.Sol. 86 (1986),394.

- 5. A.F. Craievich, D.I. Dos Santos, M.A. Aegerter T.Lours, J.Zarzycki. Anais do 7° CBECIMAT p.451-453, December 1986. Florianopolis, SC, Brazil.
- 6. D.I. Dos Santos, M.A. Aegerter, A.F. Craievich I.Lours, J.Zarzycki J. of Non-Cryst.Sol. 95 & 96 (1987),1143.
  7. T.Lours, J.Zarzycki, A.F. Craievich, D.I. Dos Santos, M.A. Aegerter. J. of Non-Cryst.Sol. 95 & 96 (1987),1151.
- 8. C.A.M. Mulder, G. Van Leeuven-Strienstra, J.G. Van Lierop, J.P. Woerdman, J.of Non-Cryst.Sol. 82 (1986),148.
- 9. C.E. Walrafen, M.S. Mokmabadi, N.C. Holmes, W.J. Nellis, S. Henning.
  J. Chem.Phys. 82,5 (1985) 2472.
- 10.F.J. Broeker, W. Heckmann, F. Fisher, M.Meilk J. Shroeder, A.Stauge. "Aerogels, Fricke Ed., Springer (Berlin)1986, p.160.
- 11.T. Woignier, J.Phalippou. J.of Non-Cryst.Sol. <u>93</u> (1987),17.
- 12. W.Ruland, J.of Applied Cryst. 4(1971)70.